STUDENT WARNING: This course syllabus is from a previous semester archive and serves only as a preparatory reference. Please use this syllabus as a reference only until the professor opens the classroom and you have access to the updated course syllabus. Please do NOT purchase any books or start any work based on this syllabus; this syllabus may NOT be the one that your individual instructor uses for a course that has not yet started. If you need to verify course textbooks, please refer to the online course description through your student portal. This syllabus is proprietary material of APUS.

American Public University System

The Ultimate Advantage is an Educated Mind

School of Science, Technology, Engineering & Math Course Number: CHEM234 Course Name: Organic Chemistry II with Lab Credit Hours: 4 Length of Course: 16 weeks Prerequisite: CHEM233 Organic Chemistry I with Lab, with a grade of a C- or higher

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Course Description

This is the second course of a two-course organic chemistry sequence that introduces students to organic reactions, reaction mechanisms, synthesis, and spectroscopy building on the foundation of Organic Chemistry I, including functional group reactivity of carbonyl compounds, aromatic compounds, organic acids and bases, polymers, and biomolecules. The laboratory component of this course is hands-on and designed for science students to learn how to make gualitative and guantitative observations about physical and chemical phenomena, make calculations, and test their own reasoning. Students will acquire skills in laboratory techniques designed to help reinforce and build upon the concepts presented in the lecture portion of the class. Students must have a safe work area available to perform laboratory activities, including working with an open flame. Students must be able to document their laboratory work using digital pictures and/or video. Students must also have room temperature storage available in order to maintain laboratory materials. Lab material for this course will only be provided once. If you need replacement lab equipment for any reason or need to retake the course later, you will need to purchase your own lab refills. This is a time and resource-intensive course. Students intending to use this course to satisfy prerequisites for pre-professional programs should verify that the Organic Chemistry with Lab course sequence meets the requirements of their intended program prior to enrollment. In order to be successful in this course, students must have successfully completed Organic Chemistry I with Lab or equivalent. NOTE: This course requires the student to purchase additional materials at substantial cost that are not covered by the book grant.

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Course Scope

This course is designed to teach the principles of organic chemistry and its laboratory to students who are science majors. It builds upon the content from Organic Chemistry I with Lab, and it will introduce organic reactions, reaction mechanisms, synthesis, and spectroscopy building on the foundation of Organic Chemistry I, including functional group reactivity of carbonyl compounds, aromatic compounds, organic acids and bases, polymers, and biomolecules, and provide a solid foundation of organic chemistry for subsequent science courses.

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Course Objectives

- CO1 Predict structure and reactivity of the following types of organic molecules: aldehydes, ketones, carboxylic acids, esters, anhydrides, amides, b-ketocarbonyl containing compounds, amines, and phenols.
- CO2 Provide curved arrow mechanisms for the following types of organic reactions: electrophilic & nucleophilic aromatic substitution, nucleophilic carbonyl addition, nucleophilic acyl substitution, carbonyl condensation, [4,2] cycloaddition, and substitution to a carbonyl.
- CO3 Predict the reactivity of the following reactive species: Grignard reagents, Wittig reagents, organolithiums, diorganocuprates, enols, enolates, aryl diazonium ions, and benzynes.
- CO4 Recognize the structure and reactivity of the following classes of bioorganic molecules: carbohydrates, amino acids, proteins, and lipids.

Course Delivery Method

This course delivered via distance learning will enable students to complete academic work in a flexible manner, completely online. Course materials and access to an online learning management system will be made available to each student. Online assignments are due by Sunday evening of the week as noted and include Discussion questions (accomplished in groups through a threaded discussion), examination, Labs, and individual assignments submitted for review by the Faculty Member). Assigned faculty will support the students throughout this course.

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Course Resources

Textbook (OER): McMurry, J. (2019). Organic chemistry. Libretexts. Lab Materials: Hands-On Labs (2020) Organic Chemistry, eScience.

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Evaluation Procedures

Reading Assignments: Syllabus, Academic Honor Pledge, LibreTexts Text: Chapters 14 - 26

Quizzes: Quizzes 1 - 6

Introduction Discussion Assignment: Introduction Discussion

Discussion Assignments: Discussions 1 - 8

Exams: Midterm Exam; Final Exam

Laboratory Assignments: Labs 1 - 6

Late Work: Minus 10% per day late

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16 – Week Course Outline and Grading

Weeks 1 & 2

Course Objectives: CO-1; CO-3

Reading/Viewing: Syllabus, Academic Honor Pledge, LibreTexts Text: Chapters 14 & 15

Discussions:	Week 1 - Introduction Discussion & Replies	2.22 %
	Week 2 - Week 2 Discussion & Replies	2.22 %
Lab:	Week 2 - Lab 1	3.75 %
Assessment:	Week 2 - Quiz 1	3.57 %

Weeks 3	8&4
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Course Objectives: CO-1; CO-3

Reading/Viewing: Syllabus, LibreTexts Text: Chapters 16 & 17

Discussions:	Week 3 - Week 3-4 Discussion	
	Week 4 - Week 3-4 Discussion & Replies	2.22 %
Lab:	Week 4 - Lab 2	3.75 %
Assessment:	Week 4 - Quiz 2	3.57 %

Weeks 5 & 6

Course Objectives: CO-1; CO-2; CO-3		
Reading/Viewing: Syllabus, LibreTexts Text: Chapters 18 & 19		
Discussions:	Week 5 - Week 5-6 Discussion	
	Week 6 - Week 5-6 Discussion & Replies	2.22 %
Lab:	Week 6 - Lab 3	3.75 %
Assessment:	Week 6 - Quiz 3	3.57 %
Weeks 7 & 8		
Course Objectives: CO-1; CO-2		
Reading/Viewing: Syllabus, LibreTexts Text: Chapter 20		
Discussions:	Week 7 - Week 7-8 Discussion	
	Week 8 - Week 7-8 Discussion & Replies	2.22 %
Lab:	Week 8 - Lab 4	3.75 %
Assessments:	Week 8 - Quiz 4	3.57 %
	Week 8 - Midterm Exam	12.5 %

Weeks 9 & 10

Course Objectives: CO-1; CO-2; CO-3		
Reading/Viewing: Syllabus, LibreTexts Text: Chapters 21 & 22		
Discussions:	Week 9 - Week 9-10 Discussion	
	Week 10 - Week 9-10 Discussion & Replies	2.22 %
Lab:	Week 10 - Lab 5	3.75 %
Assessment:	Week 10 - Quiz 5	3.57 %

Weeks 11 & 12

Course Objectives: CO-1; CO-2		
Reading/Viewi	ng: Syllabus, LibreTexts Text: Chapters 23 & 24	
Discussions:	Week 11 - Week 11-12 Discussion	
	Week 12 - Week 11-12 Discussion & Replies	2.22 %
Lab:	Week 12 - Lab 6	3.75 %
Assessment:	Week 12 - Quiz 6	3.57 %
Weeks 13 & 14		
Course Objectives: CO-3; CO-4		
Reading/Viewing: Syllabus, LibreTexts Text: Chapters 25 & 26		
Discussions:	Week 13 - Week 13-14 Discussion	
	Week 14 - Week 13-14 Discussion & Replies	2.22 %
Lab:	Week 14 - Lab 7	3.75 %
Assessment:	Week 14 - Quiz 7	3.57 %
Weeks 15 & 16	7	
Course Objectives: CO-1; CO-2; CO-3; CO-4		
Reading/Viewing: Syllabus, LibreTexts Text: Chapters Review 14-26		
Discussions:	Week 15 - Week 15-16 Discussion	
	Week 16 - Week 15-16 Discussion & Replies	2.22 %
Lab:	Week 16 - Lab 8	3.75 %
Assessments:	Week 16 - Final Exam	12.5 %
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Please see the Student Handbook to reference the University's grading scale.

Policies

Please see the <u>Student Handbook</u> to reference all University policies. Quick links to frequently asked question about policies are listed below.

Drop/Withdrawal Policy Plagiarism Policy Extension Process and Policy Disability Accommodations

Writing Expectations

All written submissions should be submitted in a font and page set-up that is readable and neat. It is recommended that students try to adhere to a consistent format, such as that described below.

- Typewritten in double-spaced format with a readable style and font and submitted inside the electronic classroom (unless classroom access is not possible and other arrangements have been approved by the professor).
- 11 or 12-point font in a style such as Arial, Helvetica or Times New Roman.

Citation and Reference Style

Assignments completed in a narrative essay or composition format must follow a widely accepted citation style, such as APA, Turabian or MLA. Please refer to the APUS Online Library for further examples, or contact the instructor with questions.

Late Assignments

Students are expected to submit classroom assignments by the posted due date and to complete the course according to the published class schedule. As adults, students, and working professionals, I understand you must manage competing demands on your time. Should you need additional time to complete an assignment, please contact me **before the due date** so we can discuss the situation and determine an acceptable resolution. Routine submission of late assignments is unacceptable and may result in points deducted from your final course grade.

<u>Netiquette</u>

Online universities promote the advancement of knowledge through positive and constructive debate – both inside and outside the classroom. Discussions on the Internet, however, can occasionally degenerate into needless insults and "flaming." Such activity and the loss of good manners are not acceptable in a university setting – basic academic rules of good behavior and proper "Netiquette" must persist. Remember that you are in a place for the rewards and excitement of learning which does not include descent to personal attacks or student attempts to stifle the Discussion of others.

• **Humor Note:** Despite the best of intentions, jokes and <u>especially</u> satire can easily get lost or taken seriously. If you feel the need for humor, you may wish to add "emoticons" to help alert your readers: ;-), :), □

Disclaimer Statement

Course content may vary from the outline to meet the needs of this particular group.

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Online Library

The Online Library is available to enrolled students and faculty from inside the electronic campus. This is your starting point for access to online books, subscription periodicals, and Web resources that are designed to support your classes and generally not available through search engines on the open Web. In addition, the Online Library provides access to special learning resources, which the University has contracted to assist with your studies. Questions can be directed to <u>librarian@apus.edu</u>.

- Charles Town Library and Inter Library Loan: The University maintains a special library with a limited number of supporting volumes, collection of our professors' publication, and services to search and borrow research books and articles from other libraries.
- **Electronic Books:** You can use the online library to uncover and download over 50,000 titles, which have been scanned and made available in electronic format.
- **Electronic Journals:** The University provides access to over 12,000 journals, which are available in electronic form and only through limited subscription services.

• **Tutor.com**: AMU and APU Civilian & Coast Guard students are eligible for 10 free hours of tutoring provided by APUS. <u>Tutor.com</u> connects you with a professional tutor online 24/7 to provide help with assignments, studying, test prep, resume writing, and more. Tutor.com is tutoring the way it was meant to be. You get expert tutoring whenever you need help, and you work one-to-one with your tutor in your online classroom on your specific problem until it is done.

Library Guide (http://apus.campusguides.com/SCIN134)

The AMU/APU Library Guides provide access to collections of trusted sites on the Open Web and licensed resources on the Deep Web. This course guide provides links to a number of sources relevant to this course, including journals, books, and web sites. Also, you can directly contact the librarian assigned to this course for assistance in locating information.

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